metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Reinout Meijboom

Department of Chemistry, University of the Free State, PO Box 339, Bloemfontein 9300, South Africa

Correspondence e-mail: meijboomr.sci@mail.uovs.ac.za

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.006 Å Disorder in main residue R factor = 0.040 wR factor = 0.094 Data-to-parameter ratio = 19.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Iodotris(tri-p-tolylphosphine)silver(I)

In the title compound, $[AgI(C_{21}H_{21}P)_3]$, important bond distances are: Ag-P = 2.5346 (9), 2.5562 (9) and 2.5617 (9) Å, and Ag-I = 2.8683 (5) Å. The Ag^I atom is in a distorted tetrahedral environment and all bond angles are close to the ideal values.

Comment

As part of our investigation of the coordination of tritolylphosphine to silver(I) (Meijboom, 2006; Meijboom & Muller, 2006, Meijboom *et al.*, 2006; Venter *et al.*, 2006), we report here the solid-state structure of the title compound, $[Ag[P(4-MeC_6H_4)_3]_3I]$, (I).



The Ag atom of (I) is coordinated by three phosphine ligands and one iodide anion, forming a distorted tetrahedral environment (Table 1). All bond angles around the Ag atom



The molecular structure of (I), showing 30% probability displacement ellipsoids. H atoms have been omitted for clarity. For the C atoms, the first digit indicates the ring number and the second digit indicates the position of the atom in the ring. Some labels have been omitted for clarity, but all rings were numbered in a similar fashion.

© 2007 International Union of Crystallography All rights reserved Received 23 November 2006 Accepted 24 November 2006 are close to the ideal values for a tetrahedral environment. Some thermal motion of ring 1 was observed but was not regarded as large enough for disordered refinement.

Experimental

Silver(I) iodide (100 mg, 0.43 mmol) was added to a solution of tri-ptolylphosphine (395 mg, 1.15 mmol) in pyridine (5 ml) and heated at 373 K for 5 min. The resulting colourless solution was cooled and the title compound crystallized on standing in quantitative yield as colourless crystals.

Crystal data

$[AgI(C_{21}H_{21}P)_3]$	V = 2802.7 (6) Å ³	
$M_r = 1147.81$	Z = 2	
Triclinic, P1	$D_x = 1.36 \text{ Mg m}^{-3}$	
a = 11.0426 (13) Å	Mo $K\alpha$ radiation	
b = 11.5665 (14) Å	$\mu = 1.03 \text{ mm}^{-1}$	
c = 23.243 (3) Å	T = 293 (2) K	
$\alpha = 99.292 (3)^{\circ}$	Block, colourless	
$\beta = 92.174 \ (2)^{\circ}$	$0.3 \times 0.26 \times 0.15 \text{ mm}$	
$\gamma = 106.196 \ (2)^{\circ}$		
Data collection		
Bruker SMART 1K CCD area-	19391 measured reflections	
detector diffractometer	13246 independent reflections	
ω scans	8623 reflections with $I > 2\sigma(I)$	

 $R_{\rm int} = 0.022$

 $w = 1/[\sigma^2(F_0^2) + (0.0327P)^2]$

+ 1.1965P] where $P = (F_0^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\rm max} = 0.57 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.54 \text{ e} \text{ Å}^{-3}$

 $\theta_{\rm max} = 28^{\circ}$

 ω scans Absorption correction: multi-scan (SADABS; Bruker, 2004) $T_{\min} = 0.747, \ T_{\max} = 0.860$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.040$ wR(F²) = 0.094 S = 1.0113246 reflections 677 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

Ag-P1	2.5346 (9)	Ag-P3	2.5617 (9)
Ag-P2	2.5562 (9)	Ag-I	2.8683 (5)
P1-Ag-P2	112.04 (3)	P1-Ag-I	102.35 (2)
P1-Ag-P3	117.65 (3)	P2-Ag-I	99.38 (2)
P2-Ag-P3	111.94 (3)	P3-Ag-I	111.51 (2)

The methyl and aromatic H atoms were placed in geometrically idealized positions (C-H = 0.92-0.98 Å) and constrained to ride on their parent atoms, with $U_{iso}(H) = 1.5U_{ea}(C)$ for methyl H or $1.2U_{ea}(C)$ for aromatic H. Methyl H atoms were placed in idealized positions and the torsion angles refined to fit the electron density. The disordered tolyl substituent (C92-C97) was refined with complementary group occupancy factors; these refined to 0.648 (17)/ 0.352 (17).

Data collection: SMART-NT (Bruker, 1999); cell refinement: SAINT-Plus (Bruker, 1999); data reduction: SAINT-Plus and XPREP (Bruker, 1999); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Brandenburg & Putz, 2005); software used to prepare material for publication: WinGX (Farrugia, 1999).

Financial assistance from the South African National Research Foundation, THRIP, the Research Fund of the University of the Free State and SASOL is gratefully acknowledged. The University of the Witwatersrand (Professor D. Levendis and Dr D. G. Billing) is thanked for the use of its diffractometer. The author is grateful to Professor A. Roodt (University of the Free State) for the freedom to pursue his own research within his group. Part of this material is based on work supported by the South African National Research Foundation (SA NRF, GUN 2038915). Opinions, findings, conclusions or recommendations expressed in this material are those of the author and do not necessarily reflect the views of the NRF.

References

- Brandenburg, K. & Putz, H. (2005). DIAMOND. Release 3.0c. Crystal Impact GbR, Bonn, Germany.
- Bruker (1999). SAINT-Plus (Version 6.02, including XPREP) and SMART-NT (Version 5.050). Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2004). SADABS. Version 2004/1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Meijboom, R. (2006). Acta Cryst. E62, m2698-m2700.
- Meijboom, R. & Muller, A. (2006). Acta Cryst. E62, m3191-m3193.
- Meijboom, R., Muller, A. & Roodt, A. (2006). Acta Cryst. E62, m2162-m2164. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Venter, G. J. S., Meijboom, R. & Roodt, A. (2006). Acta Cryst. E62, m3453m3455.